

Polymer-Protected Gold Nanoparticles for Photothermal Treatment of Ehrlich Adenocarcinoma: In Vitro and In Vivo Studies

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Photothermal therapy (PTT) is recognized as an effective tool for the treatment of cancer and it has attracted considerable attention of scientists. In this work, gold nanospheres (AuNSs) and gold nanorods (AuNRs) stabilized using poly(*N*-vinylpyrrolidone) (PVP), pristine gellan gum (PGG), and poly(2-ethyl-2-oxazoline)-grafted gellan gum (GG-g-PEtOx) are synthesized and evaluated as PTT agents in Ehrlich cancer cells. The physicochemical characteristics of these AuNSs and AuNRs, including their surface plasmon resonance absorption spectra, size, zeta potential, and aspect ratio are studied using UV-vis-spectroscopy, dynamic light scattering, zeta potential, transmission electron microscopy, and optical microscopy techniques. The polymer-protected AuNSs exhibit light-to-heat conversion, raising the temperature from 37 to 43 °C when irradiated using a visible light source. In the case of AuNSs, considerable damage to Ehrlich cancer cells is observed following irradiation and 40 days of examination. However, with regard to AuNRs, the damage to Ehrlich cancer cells is slightly lower than observed in AuNSs. In vivo experiments demonstrate that laser irradiation of tumors in mice after injecting AuNSs leads to a statistically significant decrease in tumor size as compared to those not irradiated and the control samples.

Unfortunately, due to the heterogeneous nature of cancer, which poses a significant public health challenge, there are currently no fully inclusive approaches to effectively treat this condition.^[5] The primary modalities currently employed for cancer treatment include chemotherapy, radiation therapy, immunotherapy, and surgery. These methods have become widely used in clinical practice for decades. However, cancer patients undergoing these therapies often experience significant adverse effects.^[6] For this reason, the majority of studies in cancer therapy are focused on the development of alternative therapies that can complement or even substitute the existing therapies. The goal is to improve their efficacy and minimize any potential side effects that they may have on patients. These methods ideally should selectively eliminate cancerous cells only, without damaging healthy cells.^[7,8] Among the advanced phototherapy methods for treating cancer, photothermal therapy (PTT) offers a great advantage due to its noninvasive nature and selective

therapeutic potential for different cancers. PTT has several advantages including the ability to externally irradiate tumors, which means the therapy can be applied from outside the body. PTT is also associated with limited complications, meaning it has a lower risk of side effects compared to other cancer

1. Introduction

Cancer is a multifaceted disease characterized by uncontrolled growth and spread of abnormal cells in the body and is one of the leading causes of human morbidity and mortality worldwide.^[1–4]

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therapies. Additionally, PTT offers enhanced selectivity, meaning it can specifically target cancer cells while minimizing the damage to normal cells. Another benefit of PTT is a relatively quick patient recovery, implying that they may experience a faster recuperation period after undergoing PTT. In phototherapy, certain wavelengths of light within the visible and near-infrared (NIR) resonance bands are utilized to heat photothermal agents such as nanoparticles. This leads to a localized increase in the temperature of specific tissues resulting in the elimination of malignant cells in those tissues. The higher sensitivity of cancer cells to temperature elevation results in their increased susceptibility to the effects of heat compared to normal cells. Moreover, an exposure to an external laser with adjustable dosing facilitates the selective eradication of various types of cancer cells while minimizing the damage to the surrounding healthy tissues.^[9–14]

Gold nanoparticles (AuNPs) are considered to be the foremost photothermal agents used in PTT treatment. These nanoparticles have an important attribute such as a high light-to-heat conversion efficiency, which makes them particularly effective in converting light energy into heat. In PTT, visible and NIR laser lights are employed to activate and stimulate AuNPs, which cause the nanoparticles to resonate and generate heat, thus enabling localized heating of the targeted tumor tissues. The application of AuNPs as photothermal agents provides enhanced therapeutic outcomes due to the possibility of their direct injection into the tumor, while minimizing nonspecific distribution in the body. Furthermore, these nanoparticles can be safely eliminated from the body after completing the therapy.^[15–17]

AuNPs exhibit distinctive physicochemical properties such as a possibility for localized surface plasmon resonance (LSPR). The LSPR phenomenon facilitates interactions between incident light and the electrons in the conduction band on the surface of AuNPs.^[18–20] The optical properties of AuNPs are dependent on their dimensions. The LSPR properties can be altered by modifying the shape and size of AuNPs. This enables the use of different wavelengths of light, including those in the NIR and visible spectra for achieving localized heating effects.^[21–24] Therefore, the ability to tune the LSPR properties of AuNPs through size and shape manipulation offers flexibility in designing AuNPs for specific applications that require interaction with different wavelengths of light. Visible light has limited penetration ability into biological tissues compared to NIR light. This property makes visible light more suitable for certain medical procedures that require higher precision.^[25–28]

Small spherical gold nanoparticles are easy to produce.^[12] They are less toxic than particles of large size and show improved photoconversion capability that may result from the irradiation with the visible light using standard surgical green lasers. In particular, the efficiency of light-to-heat conversion (photoconversion at 530 nm) of 14 nm gold nanospheres (AuNSs) irradiated in the visible region and its application to selectively obliterate cancer cells using breast cancer as a model was shown. Thus, the AuNSs with diameters ranging from 10 to 30 nm were considered nontoxic due to such properties, as stability, cellular uptake efficiency, and favourable clearance mechanism. They were recognized as ideal photothermal agents for biomedical applications^[29,30] and demonstrated a characteristic LSPR band around 520 nm, i.e., in the visible region of the spectrum, with an efficient light-to-heat conversion.^[26,31] As such, AuNSs, as pho-

tothermal agents, are an efficient way to induce precise heating leading to less damage to surrounding tissues, while destroying malignant thermosensitive cells.^[25,32]

Many synthetic and natural polymers can function as both reducing and capping agents to form AuNPs.^[33,34] According to literature, the efficacy of PTT with regard to the antitumor activity of AuNSs and gold nanorods (AuNRs), among other noble and transition metal nanoparticles stabilized with natural and/or synthetic polymers, has been studied with respect to melanoma,^[35] 4T1 and HeLa cells,^[36] 4T1 murine breast tumor cells,^[37] U87 MG human glioblastoma cell,^[38] and transplanted liver tumor.^[39]

In the present work, AuNSs and AuNRs stabilized using poly(*N*-vinylpyrrolidone) (PVP), pristine gellan gum (PGG), and poly(2-ethyl-2-oxazoline)-grafted gellan gum (GG-g-PEtOx) were prepared and characterized for their size and morphology using dynamic light scattering (DLS) and transmission electron microscopy (TEM). Coating and stabilization of AuNPs and AuNRs by biocompatible, nontoxic, and biodegradable polymers, such as poly(*N*-vinylpyrrolidone) and gellan, allow us to preserve the colloidal gold nanoparticles in aqueous solution for an extended period and to prepare effective photothermal agent for the treatment of cancer cells.^[40–48] Polymer-protected gold nanoparticles were produced through one-pot and growth seeding methods in aqueous solutions. These particles were observed to exhibit temperature-dependent changes and high stability over a period of 36 days, thus making them suitable for application in photothermal therapy. The protocols have been developed herein for the evaluation of the photothermal conversion ability of AuNSs and AuNRs when irradiated using visible and NIR laser light sources. The potential use of AuNSs and AuNRs in PTT treatment was tested both in vitro and in vivo against Ehrlich cancer cells. Ehrlich ascites carcinoma is a well-established murine model used for studying breast cancer.^[49,50] It represents hyperdiploid and an undifferentiated carcinoma with 100% malignancy, short life span, high transplantable capability, and rapid proliferation. Since 2010, there has been an increase in the number of publications mentioning Ehrlich tumours. This is due to the systematic alterations induced by the tumors, the sensitivity of the tumor cells to chemotherapies and the antitumor potential of synthetic and natural products.

It is known^[51] that malignant cells require large amounts of monosaccharide uptake in order to sustain their accelerated growth and division in comparison to that of healthy cells. We hypothesize that gellan gum composed of tetrasaccharide repeating units (1,3- β -D-glucose, 1,4- β -D-glucuronic acid, 1,4- β -D-glucose, and 1,4- α -L-rhamnose) may function as a “food” for cancer cells to support their accelerated growth. It is supposed that the cancer cells, feeding on tetrasaccharides, consume them and thereby “bare” gold nanoparticles that have been stabilized by the polysaccharide—gellan. Gold nanoparticles lacking a protective shell consisting of gellan will be forced to adsorb on cancer cells in order to minimize the free energy of the nanoparticles. Irradiation of these gold nanoparticles, attached to the cancerous cells, with appropriate laser light can induce local heating due to the LSPR and hyperthermia causing apoptosis of the cancer cells. In addition, GG-g-PEtOx could act as a bioadhesive^[52] or a mucus-penetrating polymer.^[53] Therefore, coating or conjugating gold particles with these polymers could facilitate the adhesion of gold nanoparticles to the mucosal

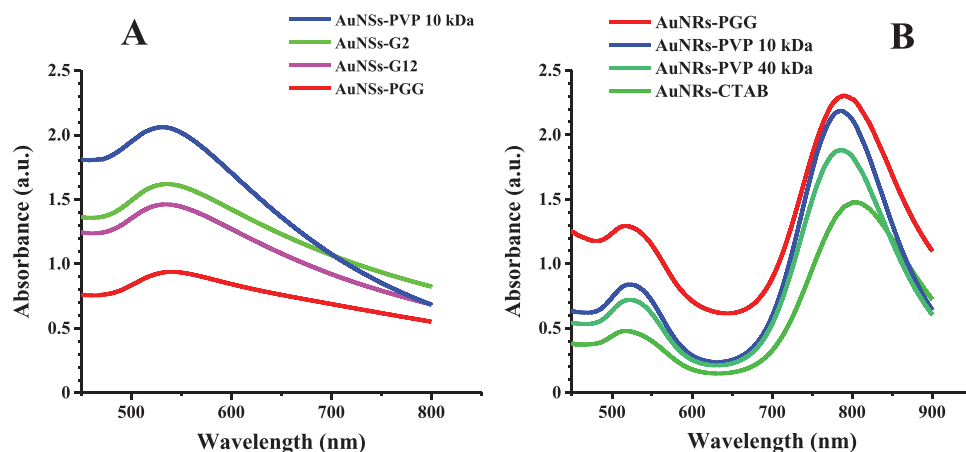


Figure 1. Representative surface plasmon resonance (SPR) absorption bands of A) spherical and B) rod-shaped gold nanoparticles stabilized using different amounts of polymers and a surfactant. Pristine gellan gum (PGG, 0.5%); G2 and G12 are poly(2-ethyl-2-oxazoline)-grafted gellan gums with different grafting densities (each at 2%); poly(*N*-vinylpyrrolidone) (PVP) with $M_n = 10$ and 40 kDa (4%), and cetyltrimethylammonium bromide (CTAB, 0.2 M). All spectra were recorded at 25 °C.

surface of tumors or their penetration into tumors. This process may prolong the efficacy of PTT by ensuring better localization of gold nanoparticles within the tumor tissue.

2. Results and Discussion

2.1. Physicochemical Characterization of Nanoparticles

Poly(*N*-vinylpyrrolidone) with $M_n = 10$ and 40 kDa, PGG, and poly(2-ethyl-2-oxazoline)-grafted gellan gums (abbreviated as G2, G3, and G12) were used as polymeric stabilizing agents of the AuNSs and AuNRs. It was expected that the surface modification of AuNPs with polymers can facilitate their cellular uptake. This enhancement is attributed to the affinity of both natural and synthetic polymers to cellular membranes.^[54,55]

UV–vis spectroscopy is an effective method for confirming the formation and stabilization of aqueous dispersions of AuNSs and AuNRs. In general, AuNSs possess a single absorption band in the visible (500–550 nm) spectral region, which is known as a surface plasmon resonance (SPR). AuNRs exhibit two major absorption bands corresponding to the transverse and longitudinal SPR bands in the visible (≈ 520 nm) and the near-infrared regions, respectively.^[56,57] The SPR spectrum is dependent on both the size and shape of gold nanoparticles. In this study, the adsorption spectra of polymer-coated AuNSs and AuNRs dispersions were recorded (Figure 1). The spectra confirmed a distinctive SPR band for AuNSs in the visible (≈ 530 nm) region (Figure 1A), while two characteristic surface plasmon bands corresponding to the transverse (≈ 520 nm, a weaker band in the visible region) and longitudinal (≈ 780 nm in NIR region) bands were observed for gold nanorods (Figure 1B).^[54,57,58]

Controlling the size of AuNPs is a critical factor in the synthesis of colloidal gold and their application in PTT. Many studies have demonstrated that the interaction between gold nanoparticles and polymers significantly influences the size, stability, and size distributions of the particles.^[59,60] Producing monodisperse polymer-coated AuNPs with smaller sizes in aqueous media can

lead to improved biocompatibility, reduced cytotoxicity, and enhanced catalytic properties.

The average hydrodynamic size and zeta potential values of the spherical and rod-shaped AuNPs synthesized and stabilized with optimally selected concentrations of synthetic and natural polymers are summarized in Tables 1 and 2, respectively. During the dialysis, the pH of aqueous solution of AuNPs decreased from 12 to 8 confirming that most of the low-molecular-weight impurities are washed out. The mean diameter and zeta potential values of AuNSs were measured by DLS before and after dialysis against deionized water. It was observed that after dialysis, the size of AuNSs decreased by 2–3 times (Table 1). Overall, most of the spherical gold nanoparticles were polydisperse. The size distributions (before and after dialysis) of spherical gold nanoparticles stabilized with different polymers determined with DLS are shown in Figure S2 (Supporting Information).

In order to estimate the nanoparticles surface charge, the zeta potential values of the different AuNPs were measured. It appears that the zeta potential values of AuNPs significantly decreased following the dialysis, resulting in a further reduction of negative charges. The average zeta potential values of AuNSs were between -42 and -10 mV depending on the polymers used and dialysis (Table 1).

AuNRs stabilized using PGG displayed a negative zeta potential (-30 ± 3 mV) value due to the presence of carboxylic groups in the macromolecular chains of the polysaccharide (Table 2). Initially, the surface of AuNRs stabilized using PVP 10 and 40 kDa was positively charged, perhaps due to the presence of some excessive counterions of cetyltrimethylammonium bromide (CTAB); however, after dialysis, AuNR's zeta potential decreased down to -7 and -6 mV.

The TEM images (Figure 2) further confirmed the DLS results regarding the size of AuNPs, and both AuNSs and AuNRs were uniformly distributed. For instance, AuNSs in all batches displayed particles with sizes less than 40 nm in diameter. AuNSs stabilized with G2 (GG-g-PtEtOx) demonstrated even smaller size with 10 ± 1 nm. The aspect ratio, i.e., length/width, for the gold

Table 1. Size and zeta potential values of polymer-protected AuNSs determined by DLS (G2, G3, and G12: poly(2-ethyl-2-oxazoline)-grafted gellan gums with different grafting densities; PDI: polydispersity index; PGG: pristine gellan gum; PVP: poly(*N*-vinylpyrrolidone). Data are expressed as mean $\pm$ standard deviation values ($n = 3$)).

Type of polymer	Concentration of polymer [% w/v]	Mean diameter [nm]				Zeta potential [mV]	
		Before dialysis	PDI	After dialysis	PDI	Before dialysis	After dialysis
PVP 10 kDa	4.0	42 $\pm$ 2	0.353	15 $\pm$ 1	0.309	−13 $\pm$ 2	−19 $\pm$ 2
PVP 40 kDa	4.0	47 $\pm$ 2	0.225	16 $\pm$ 2	0.238	−7 $\pm$ 2	−10 $\pm$ 3
PGG	0.5	88 $\pm$ 2	0.300	37 $\pm$ 2	0.415	−32 $\pm$ 2	−42 $\pm$ 3
G12	2.0	38 $\pm$ 1	0.513	13 $\pm$ 1	0.795	−21 $\pm$ 2	−33 $\pm$ 2
G3	2.0	22 $\pm$ 1	0.587	17 $\pm$ 1	0.671	−20 $\pm$ 2	−40 $\pm$ 2
G2	2.0	39 $\pm$ 1	0.496	11 $\pm$ 1	0.496	−18 $\pm$ 2	−39 $\pm$ 2

Table 2. Aspect ratio (measured using TEM) and zeta potential values of AuNRs synthesized and stabilized using PGG, G2, and PVPs (CTAB: cetyltrimethylammonium bromide; PGG: pristine gellan gum; PVP: poly(*N*-vinylpyrrolidone); n/a: not applicable. Data are presented as mean $\pm$ standard deviation ($n = 3$)).

Type of polymer	Concentration of polymer [% w/v]	Average length [nm]	Average width [nm]	Aspect ratio	ζ -potential [mV]
CTAB	n/a	42 $\pm$ 4	14 $\pm$ 2	3 $\pm$ 1	54 $\pm$ 11
PGG	0.5	34 $\pm$ 3	9 $\pm$ 2	4 $\pm$ 1	−30 $\pm$ 3
G2	2.0	53 $\pm$ 5	18 $\pm$ 3	3 $\pm$ 1	−17 $\pm$ 2
PVP 10 kDa	4.0	44 $\pm$ 13	15 $\pm$ 3	3 $\pm$ 1	−7 $\pm$ 1
PVP 40 kDa	4.0	55 $\pm$ 2	16 $\pm$ 1	4 $\pm$ 1	−6 $\pm$ 2

nanorods stabilized using PVP 10 kDa was ≈ 3 –4 in average (44 $\pm$ 13 nm lengthwise by 15 $\pm$ 3 nm in width) and the microphotographs are displayed in Figure 2 (also see Figure S3 in the Supporting Information).

2.1.1. Study of Photothermal Effects of AuNPs

The photothermal effect (i.e., light-to-heat conversion) of AuNSs was studied on porcine stomach tissues as an ex vivo tissue model. The temperature change in tissue samples was measured with an IR thermometer upon irradiation using a 530 nm visible light laser. Visible light was chosen for the current study to avoid unwanted heat stimulation, as it is generally not strongly absorbed by the bulk tissue (Figure 1A). Figure 3 shows the temperature changes in porcine stomach tissue samples with and without injection of polymer-stabilized AuNSs' dispersion as a function of irradiation time. The highest temperature recorded was 43.5 $\pm$ 0.2 $^{\circ}$ C for AuNSs–PVP 40 kDa dispersion without tissue sample when exposed to irradiation for 60 min. For the tissue sample containing AuNSs–PVP 40 kDa; PGG and G2 (GG-g-PeOx), the maximum temperature reached up to 41.1 $\pm$ 0.2, 41.0 $\pm$ 0.2, and 40.7 $\pm$ 0.2 $^{\circ}$ C, respectively, whereas the tissue sample without AuNSs was heated up to 39.0 $\pm$ 0.2 $^{\circ}$ C during 60 min of irradiation. These results clearly indicate that photothermal heating was induced by AuNSs in the tissue samples when exposed to the visible light source.

As seen from Figure 3, aqueous dispersion of AuNSs stabilized by PVP 40 kDa without tissue sample shows higher heating temperature upon irradiation compared to tissue-containing samples at identical conditions. Exact explanation of this phenomenon

is complicated because the light-to-heat conversion depends on many parameters, in particular incident laser power, light wavelength, irradiation time, concentration, and size and shape of AuNPs.^[61] In our opinion, overall increase in the temperature may be due to a collective heating effect of many nanoparticles within the excitation volume.^[62] Our results are consistent with the data of the authors^[63] indicating that the cancerous tissues can be overheated above the physiological level (typically 39–45 $^{\circ}$ C). In our case, the heating reaches up to ≈ 40 –43 $^{\circ}$ C ($\Delta T \approx 3$ –6 $^{\circ}$ C). The temperature profiles of colloidal AuNPs' solutions (not stabilized by polymers) with different particle sizes were studied.^[64] The temperature of the AuNPs solution increased exponentially upon laser illumination ($\lambda = 532$ nm), reached the equilibrium after ≈ 1200 s, and then returned to the ambient value after discontinuing irradiation. Upon irradiation the system is averagely heated up to 31.5 $^{\circ}$ C ($\Delta T \approx 6.5$ $^{\circ}$ C). Our observations are in good agreement with these results. The photothermal properties of gold nanoparticles, in addition to laser power, light wavelength, irradiation time, concentration, and size and shape, depend on various experimental factors, such as stirring, data recording and analysis, and the effective mass of the system. Moreover, in the presence of natural tissue, other molecules adsorb 530 nm light (e.g., hemoglobin and myoglobin), which makes the irradiation less effective. However, 530 nm light corresponds to the maximum of SPR and makes the photothermal phenomenon more effective. However, the 530 nm light is the most appropriate for laboratory tests and basic studies; it is less suitable for work with natural tissues. Short wavelengths are often adsorbed by natural molecules, e.g., hemoglobin and myoglobin have absorbance maxima about this value. This decreases the effectiveness of the photothermal treatment, and one has to

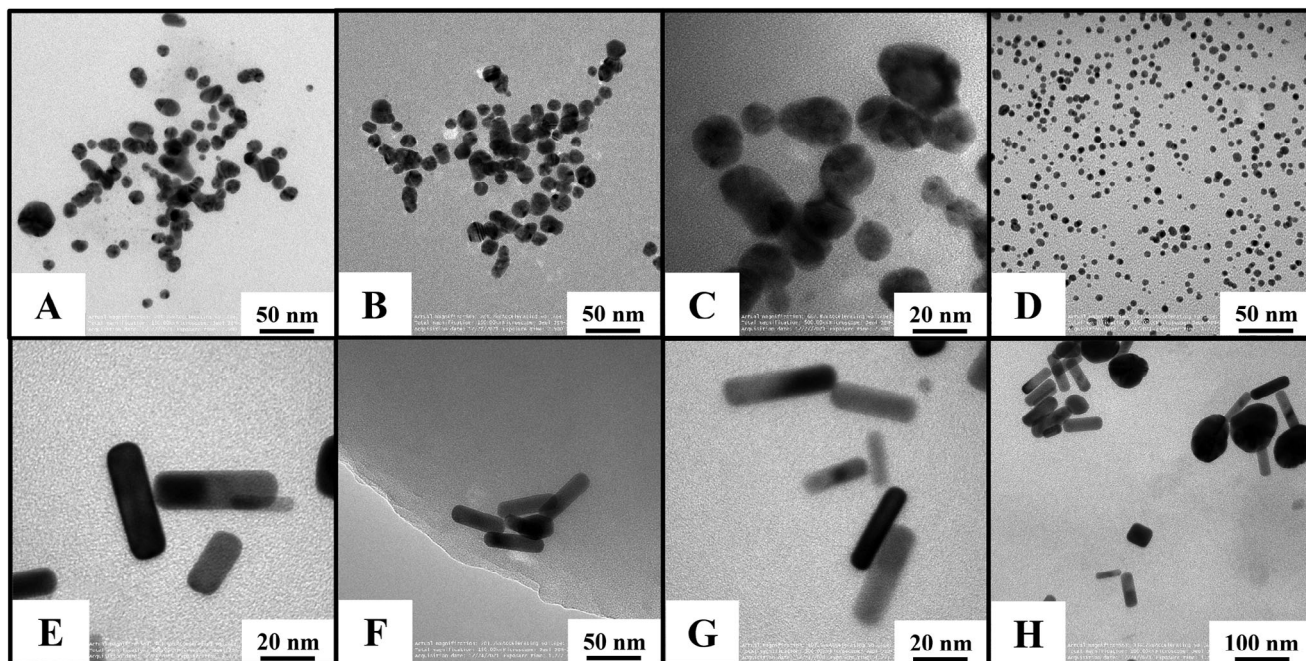


Figure 2. TEM images of AuNSs and AuNRs stabilized using A,E) poly(*N*-vinylpyrrolidone) with $M_n = 10$ kDa; B,F) poly(*N*-vinylpyrrolidone) with M_n 40 kDa; C,G) pristine gellan gum; D) G2-poly(2-ethyl-2-oxazoline)-grafted gellan gum (for spherical AuNPs), and H) cetyltrimethylammonium bromide without polymer (for AuNRs).

work with the light of longer wavelengths. In our case, we used 780 nm light, which guaranteed that the studied photothermal effect originates from the light absorbance.

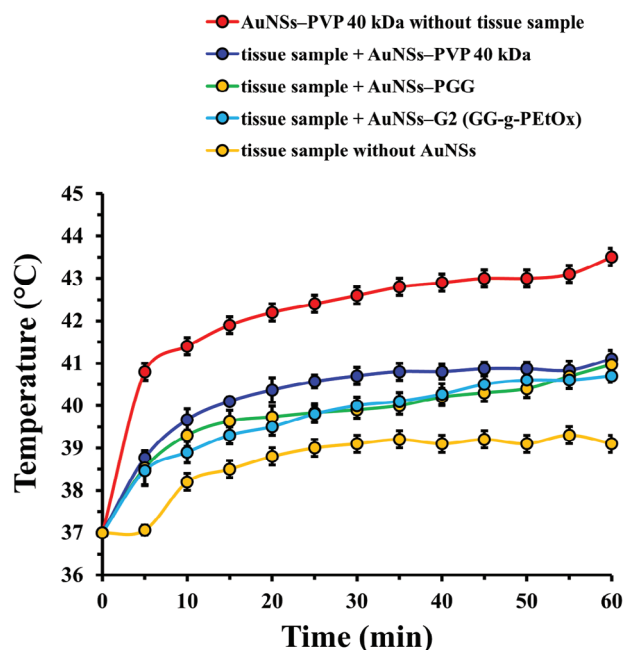


Figure 3. Time-dependent heating curves for AuNSs upon irradiation using a 530 nm visible light source. AuNSs: gold nanospheres; PGG: pristine gellan gum; G2: poly(2-ethyl-2-oxazoline)-grafted gellan gum; PVP 40 kDa: poly(*N*-vinylpyrrolidone) with $M_n = 40$ kDa.

2.1.2. Efficacy of PTT Treatment in Ehrlich Cancer Cells

Ehrlich cancer cells were used to investigate the efficacy of AuNP-based PTT treatment under the visible light irradiation. Ehrlich tumor is a well-established murine tumor model frequently employed in many cancer researches, both solid and ascitic forms. Classified as a carcinoma, it originates in the epithelial tissue of the skin or lining of internal organs. This tumor was initially derived from a spontaneous mammary adenocarcinoma in a mouse and has since been propagated in various strains of mice. Ehrlich tumor cells exhibit rapid proliferation and are characterized by their ability to induce ascites fluid accumulation in the peritoneal cavity when injected intraperitoneally. This tumor model is commonly employed to study tumor biology, tumor immunology, anticancer drug screening, and evaluating therapeutic interventions. Such attributes as high growth rate, good reproducibility, metastatic potential, and relative stability in morphological and biological characteristics were the reason to use Ehrlich cancer cells in this study.^[65–69] In order to assess the potential effect of bio-nano interactions, in vitro experiments were performed using a Hanks' balanced salt solution (HBSS) buffer solution. First, Ehrlich cancer cells were dispersed in a buffer solution at a ratio of 1:9. Then, 1 mL of the suspension of extracted Ehrlich cancer cells was mixed with 1 mL of either polymer-stabilized AuNPs or AuNRs for 5 min. The suspension was irradiated with visible (530 nm) and NIR (780 nm) laser lights for 60 min at intensities of 10 and 90 mW, respectively. Micro-photographs of the cells were then acquired from randomly selected ten points in a Goryaev chamber and Ehrlich tumor cells in the visible area of the microscope were counted (Figure 4). Interestingly, the number of cells was considerably reduced in the presence of AuNSs

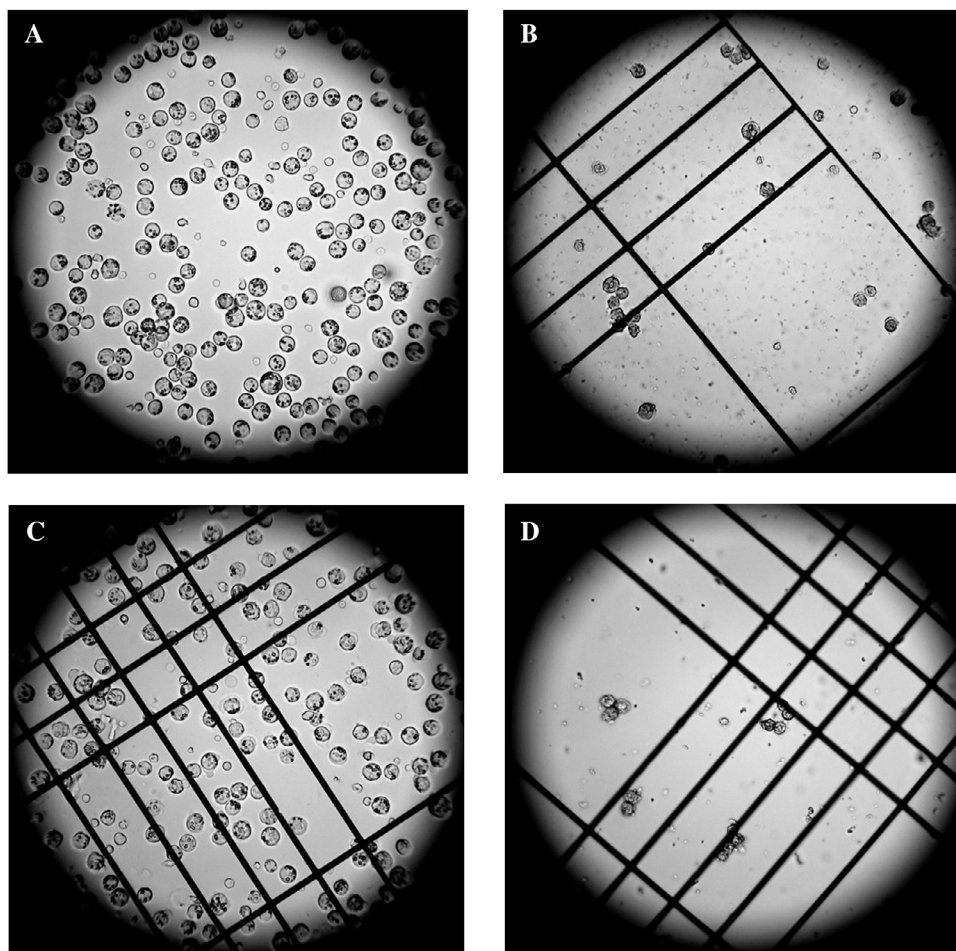


Figure 4. Micro-photographs of a suspension of Ehrlich cancer cells irradiated with visible (530 nm) light: A) before irradiation and without AuNSs; B) in the presence of AuNSs–PVP 40 kDa with no irradiation (in 40 days). C,D) After irradiation and in 40 days: C) without AuNSs, D) in the presence of AuNS–PVP 40 kDa. PVP 40 kDa: poly(*N*-vinylpyrrolidone) with $M_n = 40$ kDa.

stabilized with PVP 40 kDa even without irradiation after 40 days of examination (Figure 4B). An explanation of this phenomenon requires further experiments. No significant cell damage was observed in control experiments without AuNSs, indicating that the membranes were intact and that the cells have not been affected very much during 60 min of laser irradiation. However, due to an increase in the temperature of the medium up to 39 °C, when the samples were irradiated with either a visible or NIR light source for 60 min and further inspection during the experimental period, there was some reduction in the number of cells recorded in 40 days. As such there was no statistically significant difference between the numbers recorded in control samples with and without irradiation.

Irradiating cancer cells using visible light at 530 nm in the presence of spherical AuNPs stabilized with polymers and without yielded good results. Figure 5 displays the reduction in the number of cancer cells counted from micro-photographs taken from randomly selected ten points (numerical values are shown in Table S1 in the Supporting Information). For instance, the number of cancer cells counted after 30–40 days decreased by 10–40 times in comparison with the control samples ($p < 0.0001$). It should be noted that the number of cells in the presence of

AuNSs stabilized with polymers was reduced even without irradiation. Probably the gold nanoparticles retard the growth of cancer cells; however, irradiation enhances this process. The true mechanism of this phenomenon is not well understood and will be clarified in subsequent experiments.

Experimental results showing the effects of irradiation on the cancer cells using NIR light at 780 nm in the presence of AuNRs stabilized with polymers and control samples are illustrated in Figure 6 (numerical values are presented in Table S2 in the Supporting Information). The results are presented after 40 days of examination. The number of damaged cancer cells following irradiation increased markedly after 30–40 days of inspection; particularly, there was a statistically significant difference between the numbers recorded for polymer-stabilized AuNRs and control samples without gold nanorods ($p < 0.0001$). At the same time, no statistically significant difference was observed between the samples of AuNRs stabilized with polymers followed by irradiation and in 40 days of observation, demonstrating a similar PTT effect. Numerous studies have shown that gold nanorods with an appropriate dimension (28 × 8 nm) are the most effective for PTT compared to other types of gold nanoparticles, such as nanospheres and nanoshells.^[9,70] Despite the AuNRs having

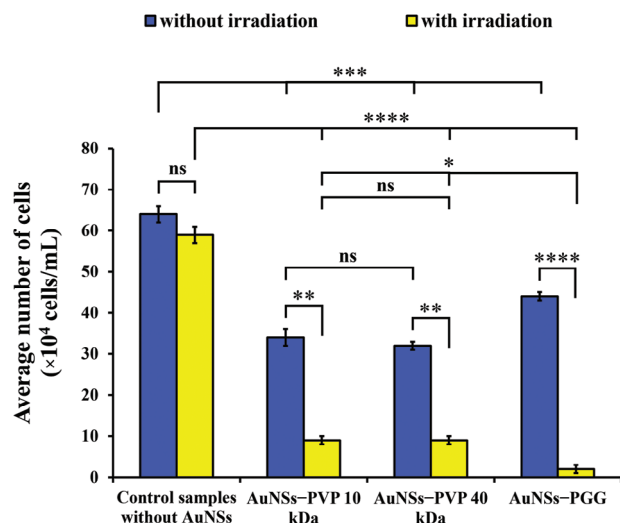


Figure 5. Number of cancer cells in control samples and in the presence of polymer-stabilized AuNSs when exposed to the visible light source at 530 nm and without irradiation. The results are presented after 40 days of examination. Data are expressed as mean $\pm$ standard error of the mean (SEM) of triplicate with $n = 10$ measurements in each. Statistically significant differences are given as **** = $p < 0.0001$; *** = $p < 0.001$; ** = $p < 0.01$; and * = $p < 0.05$; ns denotes no significance. AuNSs: spherical gold nanoparticles; PGG: pristine gellan gum; PVP 10 and 40 kDa: poly(*N*-vinylpyrrolidone) with $M_n = 10$ and 40 kDa, respectively.

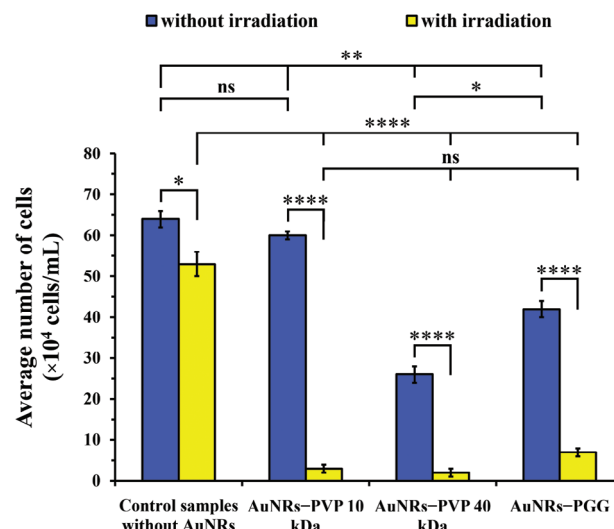


Figure 6. Number of cancer cells in control samples and in the presence of polymer-stabilized AuNRs when exposed to NIR light irradiation at 780 nm and without irradiation. The results are presented after 40 days of examination. Data are expressed as mean $\pm$ SEM of triplicate with $n = 10$ measurements in each. Statistically significant differences are shown as **** = $p < 0.0001$; ** = $p < 0.01$; * = $p < 0.05$; ns denotes no significance. AuNRs: gold nanorods; PGG: pristine gellan gum; PVP 10 and 40 kDa: poly(*N*-vinylpyrrolidone) with $M_n = 10$ and 40 kDa, respectively.

slightly larger dimensions in size, as reported in the present work (Table 2), overall, polymer-coated AuNRs demonstrated good efficacy in PTT treatment in vitro and showed a comparable PTT effect as to spherical AuNPs.

In our experiments, the photothermal conversion efficiency of AuNPs irradiated with 530 and 780 nm laser sources was not evaluated. As seen from Figure 1, for polymer-stabilized AuNSs and AuNRs, the adsorption maximums are in the range of 535 ± 5 nm and close to ≈ 800 nm. The physiotherapeutic Lasmik laser apparatus used for irradiation had only two laser sources with wavelengths of 530 and 780 nm. In case of AuNSs, the absorption band at $\lambda \approx 535 \pm 5$ nm more or less corresponds to the laser light at 530 nm. But the adsorption peaks of AuNRs at $\lambda \approx 800$ nm considerably deviate from the laser light at 780 nm. Probably this is the reason of lower efficiency of AuNRs in light-to-heat conversion compared to AuNSs. The photothermal conversion efficiency of different shapes of AuNPs under laser irradiation has been reported.^[71–74] In particular, the photothermal conversion efficiency of AuNSs and AuNRs was evaluated at different irradiation intensities of NIR broadband (754–816 nm) and NIR laser (808 nm) irradiation.^[75] It was shown that the photothermal conversion efficiency of AuNSs and AuNRs is comparable. But the spherical gold nanoparticles might be preferable for hyperthermia applications with a higher accumulation rate within the tumor sites.

2.1.3. In Vivo Study of the Photothermal Effect of AuNPs

As the in vitro experiments demonstrated successful photothermal destruction of cancer cells in the presence of polymer-stabilized AuNSs under irradiation with visible light, subsequent

in vivo studies were conducted to evaluate the therapeutic efficacy of this approach further. Tumor development was induced by injecting cancer cells into the left flanks of CD-1 mice subcutaneously (see Figure S4A in the Supporting Information), after which the tumors were allowed to grow for a period of 10 days. This is consistent with the results of the authors^[68] because the death of the animal with Ehrlich ascites carcinoma occurs between 10 and 14 days after cell inoculation depending on I) the concentration of inoculated cells, II) the amount of fluid in the peritoneal cavity, which causes abdominal pressure and compression of the organs, and III) the number of passages, in which repeated transplantation increases malignancy and tumor proliferation. Before each exposure, the linear dimensions of the tumors in animals were measured using a caliper (see Figure S4B in the Supporting Information). The volumes of the tumors were then calculated using the following equation

$$V = \frac{4}{3} \pi R^3 \quad (1)$$

where R is the radius of the tumor, and π has a value equal to 3.14.

Laser irradiation of tumor-bearing mice injected with AuNSs stabilized with PVP 40 kDa demonstrated a remarkable PTT effect (Figure 7). This result shows that further tumor growth is successfully inhibited by AuNSs–PVP injection in combination with a visible light laser irradiation.

Three days following the last exposure (see Figure S4C in the Supporting Information), the animals were humanely sacrificed, and the tumors were removed and weighed (Figure 8; Figure S4D, Supporting Information). The average tumor weight in the control group was 1.00 ± 0.30 g; in group 2 with AuNSs–PVP

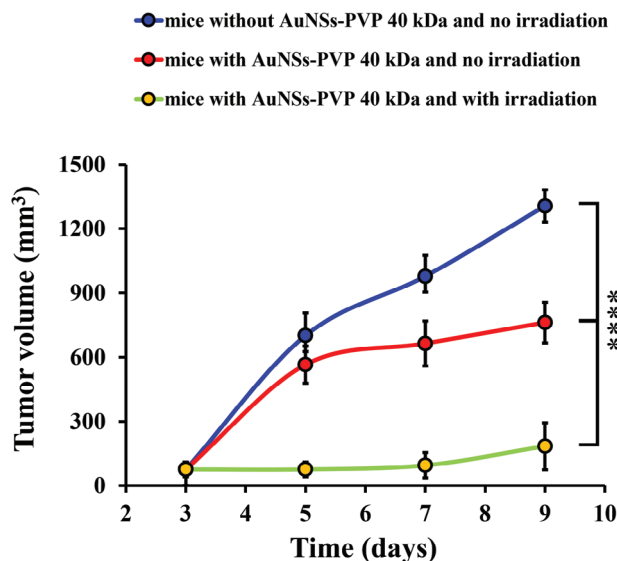


Figure 7. Tumor growth in each treatment group as monitored for 9 days. Data are expressed as mean $\pm$ standard deviation (SD) values ($n = 5$). Statistically significant difference is represented as **** = $p < 0.0001$. PVP 40 kDa: poly(*N*-vinylpyrrolidone) with $M_n = 40$ kDa.

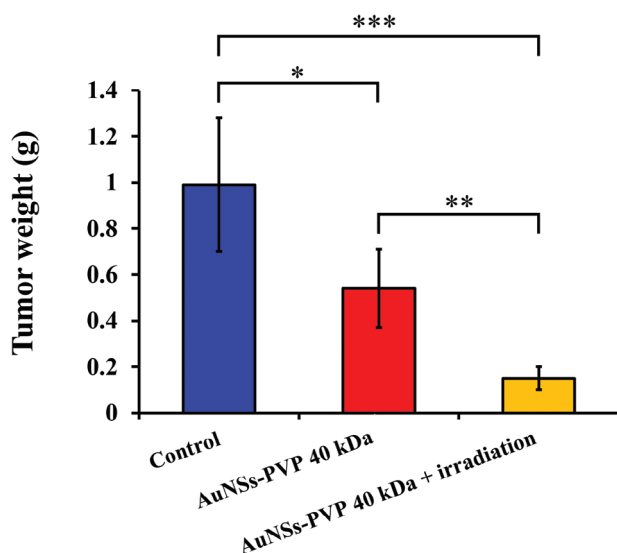


Figure 8. Tumor weight in different groups of animals measured when the animals were sacrificed. Data are produced as mean $\pm$ SD ($n = 5$). Statistically significant differences are represented as *** = $p < 0.001$; ** = $p < 0.01$; * = $p < 0.05$. PVP 40 kDa: poly(*N*-vinylpyrrolidone) with $M_n = 40$ kDa.

40 kDa it was 0.54 ± 0.20 g; and in group 3 with AuNSs-PVP 40 kDa and irradiation it was 0.15 ± 0.04 g. Figure 8 shows the comparison between tumor weight in different groups of animals measured when the animals were sacrificed.

The tumor growth inhibition coefficient (TGIC) was calculated using the following equation

$$\text{TGIC} = \frac{V_k - V_0}{V_k} \times 100\% \quad (2)$$

Table 3. Inhibition of tumor growth with intratumoral injection of AuNSs-PVP (40 kDa) before and after irradiation at $\lambda = 633$ nm (PVP 40 kDa: poly(*N*-vinylpyrrolidone) with $M_n = 40$ kDa).

Samples	Inhibition [%]			
	Tumor volume			Tumor weight
	3 days	5 days	7 days	9 days
AuNSs-PVP 40 kDa without irradiation	19 ± 10	32 ± 12	42 ± 5	43 ± 23
AuNSs-PVP 40 kDa with irradiation	86 ± 8	90 ± 4	90 ± 6	85 ± 3

where V_k is the volume (or weight) of the tumor in the control group of animals, and V_0 is the volume (or weight) of the tumor in each experimental group of animals.

The comparison of the sizes of tumors from different groups of mice was performed and the results are summarized in Table 3. The visual inspection of tumor sizes is depicted in Figure S5 (Supporting Information). Samples with AuNSs-PVP 40 kDa and without irradiation exhibited a reduced tumor growth after 3 days by $19 \pm 10\%$, which was an effect that gradually increased after 7 days, reaching $42 \pm 5\%$. Samples with AuNSs-PVP 40 kDa and with irradiation showed a greater effect with up to $90 \pm 6\%$ reduction in the tumor size after 7 days. In 9 days, TGIC calculated using the tumor weight in samples with AuNSs-PVP 40 kDa and without irradiation was $43 \pm 23\%$, and in samples with AuNSs-PVP 40 kDa plus irradiation reached $85 \pm 3\%$. Based on these results, it can be concluded that the spherical AuNSs stabilized using PVP 40 kDa could be considered as suitable candidates to inhibit Ehrlich tumor growth and could potentially be used in PTT treatment.

3. Conclusion

Spherical and rod-like gold nanoparticles protected with poly(*N*-vinylpyrrolidone), pristine gellan gum, and poly(2-ethyl-2-oxazoline)-grafted gellan gum were prepared and characterized in this study. The gold nanoparticles exhibited the presence of characteristic SPR bands. The nanoparticles were analyzed for their applicability as PTT agents with respect to Ehrlich cancer cells when exposed to the visible light source. Following the dialysis, the average hydrodynamic size of AuNPs reduced by ≈ 2 – 3 times and the zeta potential decreased by ≈ 1.5 – 2 times, indicating that the AuNPs are suitable for PTT. Experiments with ex vivo porcine stomach tissues containing AuNSs were performed to determine the photothermal effect of nanoparticles when exposed to the visible light source. In the course of laser irradiation of the tissue at 530 nm, the highest temperature recorded was 43 ± 0.5 °C for AuNSs stabilized using PVP 40 kDa. In vitro experiments demonstrated a similar PTT effect for Ehrlich cancer cells containing polymer-protected AuNPs upon irradiation both at 530 and 780 nm. After 40 days of examination, the number of Ehrlich cancer cells decreased by 10–40 times in comparison with the control samples. In vivo experiments in mice revealed that injection of AuNSs-PVP 40 kDa followed by irradiation with visible light considerably decreased the size of tumors, indicating that polymer-stabilized gold nanoparticles could potentially

be used in the PTT treatment of Ehrlich tumors. The AuNPs developed and stabilized with polymers in this work might potentially be considered as a platform for the PTT treatment of not only Ehrlich tumors, but also other type of carcinoma. In near perspectives our study may be related to light-to-heat conversion efficiency evaluation of AuNSs and AuNRs, toxicological experiments, study the bioadhesive properties polymer-protected AuNPs to improve the adhesion to cancer cell, modification of the surface of AuNPs with poly(ethylene glycol) to enhance the cellular uptake, conjugation of AuNPs with the anticancer drug doxorubicin, etc. In future perspectives the “green” synthesis using the natural substances (for instance, bacterium, fungi, and plants) may contribute to reducing and stabilizing agents for the synthesis of AuNPs and enhance their medical properties such as antimicrobial and anticancer activity.

4. Experimental Section

Materials: A standard aqueous solution of tetrachloroauric acid (HAuCl_4) with a concentration of 100 mg mL^{-1} , CTAB (99%), sodium borohydride (NaBH_4 , 98.5%), ascorbic acid, PVP with $M_n = 10$ and 40 kDa , and HBSS buffer were purchased from Sigma–Aldrich (Germany). Gellan gum with $M_w = 500 \text{ kDa}$ was purchased from Zhejiang DSM Zhongken Biotechnology Co., Ltd. (China). Poly(2-ethyl-2-oxazoline)-grafted gellan gum was kindly provided by the authors.^[70] Depending on the grafting density, the GG-g-PETox samples were abbreviated as G2, G3, and G12. Grafting density was given as the number of repeating units, where each gellan gum consists of four sugar units, per one PETox grafted chain. Thus, on average, every 12th repeating unit (equivalent to 12×4 sugar units) in G12 copolymer contains one PETox grafted chain. Potassium hydroxide, silver nitrate, and all other chemicals were of analytical grade and used as received.

Synthesis of AuNSs and AuNRs: Spherical AuNPs (or AuNSs) stabilized by PVP, PGG, and GG-g-PETox were prepared using a “one-pot” synthetic method as described previously.^[56] Briefly, a mixture consisting of polymer solutions with different concentrations (either 4% PVP; 0.5% PGG; or 2% of each G12, G3, and G2), 5 mL of HAuCl_4 (100 mg mL^{-1}), and 4 mL of 0.5 M KOH was combined, agitated, and heated up to 100°C for 3–5 min in an Anton Paar Monowave 50 microwave reactor (Graz, Austria) equipped with a temperature and time controller. Consequently, tinted solutions, varying in color from yellow to dark red or purple, were produced due to the formation of AuNSs. The initial solution with pH 12 decreased down to pH 8 following the dialysis (cellulose membrane with a molecular weight cut-off 12–14 kDa) against deionized water.

In order to prepare AuNRs, the seed-mediated growth technique was employed.^[56] Initially, a solution comprising 5 mL of 0.2 M CTAB was combined with 5 mL of 0.5 mM HAuCl_4 and stirred. Subsequently, 0.6 mL of cold 0.01 M NaBH_4 was introduced, resulting in the formation of a brownish-yellow solution containing AuNSs. Concurrently, a mixture composed of CTAB (0.2 M; 30 mL), AgNO_3 (4 mM; 1.5 mL), and HAuCl_4 (1 mM; 30 mL) was gently mixed, and 78.8 M (0.42 mL) ascorbic acid was added. Ascorbic acid was acting as a mild reducing agent, causing the color of the growth solution to change from dark yellow to colorless. Finally, the first seed solution (72 μL) was added to the second growth solution, and the mixture was allowed to incubate at 30°C overnight. Consequently, a crimson solution containing AuNRs was generated. To eliminate byproducts and CTAB from the AuNRs, the solution underwent centrifugation at 10 650 rpm for 30 min using an Eppendorf 5810R centrifuge (Tuttingen, Germany). The resulting precipitate was re-dispersed by adding 3 mL of deionized water and centrifuged again at 10 650 rpm for 15 min. After undergoing washing procedure thrice, the AuNRs were re-dispersed and stabilized in 5 mL of designated polymer solutions and then dialyzed using a cellulose membrane (molecular weight cut-off 12–14 kDa) against deionized water.

Characterization: Absorption spectra of AuNSs and AuNRs were recorded using a Specord 210 plus BU UV/Vis-spectrophotometer (Jena, Germany). The mean hydrodynamic size of gold nanoparticles in solution, their polydispersity index (PDI) and zeta potential values were determined using DLS and electrophoretic measurements with a Malvern Zetasizer Nano ZS90 (Malvern Instruments, UK) at 25°C . Gold nanoparticles were imaged using a JEOL JEM-1400Plus (JOEL Ltd., Japan) transmission electron microscope operated at an acceleration voltage of 120 kV. TEM grids were prepared by placing 10 μL of the diluted sample solutions on a carbon-coated copper grid and evaporating the solution at room temperature completely. The concentrations of AuNS and AuNR solutions stabilized with the polymers were quantified using an Agilent 7500 ICP-MS inductively coupled plasma mass spectrometer (Agilent Technologies, USA).

Study of Ex Vivo Photothermal Effect of AuNSs Induced by Visible Light Irradiation: The porcine stomach tissue was used as an ex vivo tissue model to study ex vivo photothermal effect of AuNSs when exposed to irradiation. Porcine stomach tissues were received from Altyn-Orda Abattoirs (Almaty, Kazakhstan) immediately after animal slaughter, carefully packed, transported to the laboratory in cold plastic containers, and used within 24 h of retrieval. Tissue samples (cut into $\approx 3 \times 3 \text{ cm}$) were placed in Petri dishes, and with the help of a marker pen, an area to be exposed to irradiation was highlighted. Using a sterile syringe, 0.5 mL of polymer-coated AuNSs' dispersion was injected interstitially in the porcine stomach tissue. The concentration of AuNSs in the colloidal solution was determined to be $\approx 45 \mu\text{g mL}^{-1}$. Irradiation of AuNSs was carried out using a physiotherapeutic Lasmik laser apparatus (Lazmik Ltd., Russia) (Figure S1A, Supporting Information). Matrix light-emitting diode (LED) heads operated at 530 and 780 nm wavelengths were used. Experiments with irradiation were performed within an incubator at 37°C with a relative humidity of 80% (Figure S1B, Supporting Information). The temperature was measured every 5 min using a VT04 visual infrared thermometer (Fluke VT02 Visual IR Thermometer, Everett, WA, USA). The irradiation was carried out for 60 min. All experiments were performed in triplicate and the mean $\pm$ SD values were calculated.

Irradiation of Ehrlich Cancer cells: Ehrlich cells were received from the cell culture collection of the Kazakh Research Institute of Oncology and Radiology (Almaty, Kazakhstan). The ascites' form of a transplanted Ehrlich tumor strain was used to study the photothermal effect of AuNPs on cancer cells. The tumor cells were transplanted intraperitoneally into CD-1 mice. The material was extracted on the 10th day after transplantation. The concentration of cells was determined to be 100 million mL^{-1} . In this experiment, a suspension of biomaterial was prepared from 1 mL suspension of Ehrlich tumor cells, and 9 mL of HBSS buffer solution (Hanks culture medium). Next, 1 mL of the suspension was aspirated and then mixed with 1 mL of either AuNPs ($\approx 45 \mu\text{g mL}^{-1}$) or AuNRs ($\approx 179 \mu\text{g mL}^{-1}$) solutions. The resulting mixture was subsequently irradiated for 60 min using matrix LED heads operated at $\lambda = 530$ and 780 nm. The cover glass was placed on the grid of a Goryaev chamber. Then, 5–10 μL of cell suspension was transferred to the edge of the cover glass in the chamber. Afterward, micro-photographs of the cells were taken at 40 $\times$ magnification using a Nikon Eclipse LV150N microscope (Tokyo, Japan), equipped with a Motic digital camera, and the cells were counted in Goryaev chamber manually before and after irradiation. The average number of cells in one square of grids and in 1 mL of suspension was calculated, considering the area of the square. The cells were maintained in an incubator at 37°C and in the dark throughout the experimental period. In order to evaluate the long-term photothermal effect of polymer-coated AuNSs on Ehrlich cancer cells, the cells' death was observed over a 40 day period. All measurements were conducted in triplicate and the mean $\pm$ SEM ($n = 10$) values were calculated and then evaluated statistically.

Study of In Vivo Photothermal Effect of AuNSs Induced by Visible Light Irradiation: All animal experiments were conducted in compliance with the protocol approved by the local ethical committee (LEC) at the Kazakh Research Institute of Oncology and Radiology (Protocol No.5-2021, 10 December 2021).

For the in vivo experiments, 18 CD-1 male mice of 23–25 g weight and 60 days of age were selected. Tumor cell transplantation into the animals

was accomplished through subcutaneous injection of 5×10^6 tumor cells. The experiments commenced 10 days later when the tumors reached a size of about 4–5 mm in diameter. However, throughout the experiment, some mice developed purulence and so the other three mice were excluded from the experiment due to this condition. The mice were divided into three groups: group 1 consisted of five control animals, with no AuNSs solution injected and no exposure to irradiation; group 2 included five animals with AuNSs' solution injected, but no irradiation treatment; and group 3 involved five animals with AuNSs solution injected and with irradiation treatment.

Prior to commencing each experiment, hair from the surface of the skin covering the tumor was removed, then intratumoral injection of 50 μL of AuNSs–PVP (40 kDa) was administered. The concentration of AuNSs in the suspension was $\approx 45 \mu\text{g mL}^{-1}$. After 25 min, the tumor node was subjected to a daily laser exposure for 30 min over a period of 7 days. Taking into account 10 days of inoculation and 7 days of laser exposure the total experimental time duration was 17 days. According to the protocols reported in ref. [76], the experiment associated with an increase in the tumor weight was conducted for 20 days until the animals were sacrificed. A Polaron LG-75 helium–neon laser at a wavelength of 633 nm and an emissive power of 25 mW (Kyiv, Ukraine) was used in the experiments. Throughout the experiment, the tumor size was measured daily using a caliper. Following the internationally recognized guidelines established by the Council for International Organizations of Medical Sciences (CIOMS) for biomedical research involving animals, the mice were euthanized prior to surgical removal of tumor tissues. The tumors were extracted from the peritoneum by dissection with a subsequent weighing on the ninth day.

Statistical Analysis: The data were presented as mean values $\pm$ SD and/or SEM and assessed for significance using a one-way and two-way analysis of variance (ANOVA) followed by Bonferroni posthoc test, where $p < 0.05$ was considered as the statistical significance difference (GraphPad Prism software, version 7.0; San Diego, CA, USA).

Supporting Information

Supporting Information is available from the Wiley Online Library or from the author.

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Conflict of Interest

The authors declare no conflict of interest.

Data Availability Statement

Research data are not shared.

Keywords

Ehrlich cancer cells, gold nanoparticles, gold nanorods, gold nanospheres, photothermal therapy (PTT), surface plasmon resonance (SPR)

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